EXTRACTION, ISOLATION AND EVALUATION TRIGONELLA FOENUM-GRAECUM AS MUCOADHESIVE AGENT FOR NASAL GEL DRUG DELIVERY

Dharmendra Kumar*, Aditi Singhal, Sumedha Bansal, S.K Gupta
Department of Pharmaceutical Technology, Meerut Institute of Engineering and Technology, Meerut, Uttar Pradesh, India, 250005

ABSTRACT
The purpose of the study is to formulate and evaluate a new, cheap and effective natural mucoadhesive agent that can be used as an effective alternative for traditional mucoadhesive agent. The study procedure involved extraction of mucoadhesive agent from the Trigonellafoenumgraecum (fenugreek) seeds, solubility testing of the mucilage obtained, phytochemical testing, determination of swelling index, preparation of nasal gel, measurement of viscosity. The study showed that the extraction of fenugreek seeds had 36% w/w of mucoadhesive agent. The natural mucoadhesive agent was soluble in hot water and cold water. The swelling index was found to be 160%. Fenugreek seeds produce high viscosity mucilage at low concentration levels.

Keywords: TrigonellaFoenumGraecum, Mucoadhesive Agent, Nasal Gel Drug Delivery,

*Corresponding Author:
Dharmendra Kumar
Department of Pharmaceutical Technology,
Meerut Institute of Engineering and Technology,
Meerut, Uttar Pradesh, India, 250005
Email: rvnimiet@gmail.com
INTRODUCTION

*Trigonella foenum-graecum* L. (Fenugreek) belongs to the Leguminosae family. It is an annual grassy plant. Fenugreek seeds have been widely used in food as a flavour component and seasoning.

In clinical, Fenugreek seeds are reported to have glucose and lipid-lowering properties. Phytochemical studies on *T. foenum-graecum* revealed that carbohydrates and mucilages (mainly galactomannans), proteins, fixed oils, flavonoids and saponins were the main components of the seeds.

Fenugreek is known to have several pharmacological effects such as hypoglycemic, hypocholesterolemic, antioxidant, and appetite stimulation. Furthermore, this plant has gastroprotective activity and histopathological examination of liver and brain has revealed that, aqueous extract of fenugreek seeds offer a significant protection against ethanol toxicity.

Fenugreek has been used in treating colic flatulence, dysentery, diarrhoea, dyspepsia with loss of appetite, chronic cough, dropsy, enlargement of liver and spleen, rickets, gout, and diabetes. It is also used as gastro protective, antiurolithiatic, diuretic, antidandruff agent, Anti-inflammatory agent and as antioxidant. The seed is stated to be a tonic. It also is used in post-natal care and to increase lactation in nursing mothers. Fenugreek seeds contain a high percentage of mucilage.

Since Fenugreek seeds produce high viscosity mucilage at low concentration levels, the objective of the present investigation was to evaluate the mucoadhesive effects of this mucilage in nasal gel.

MATERIALS AND METHODS

Chemicals
Acetone, and other required materials provide by department of pharmaceutics MIET Meerut.

Plant material

The seeds of Fenugreek were purchased from local market, Ghanghuali, Super Noida, India. The seeds were identified and authenticated by MIET Meerut.

METHODS

Extraction and Isolation of Mucoadhesive Material [7]
Fenugreek seeds (250 g) were soaked in double distilled water at room temperature and then boiled with sufficient amount of double distilled water under stirring condition in a water bath until slurry was prepared. Then the slurry was cooled and kept in refrigerator overnight to settle out un-
dissolved materials. The upper clear solution was decanted off and centrifuged at 1000 rpm for 30 minutes. The supernatant was separated and concentrated at 50-55°C on a water bath to a third of its original volume. Solution was cooled down to room temperature and was poured into thrice volume of acetone by continuous stirring. The precipitate was washed repeatedly with acetone and dried.

**Determination of percentage yield**
\[ \% \text{ Yield} = \frac{\text{Practical Yield} \times 100}{\text{Theoretical Yield}} \]

Physico-chemical characterization of mucilage [8-10]. The separated mucilage was evaluated for solubility, swelling index, loss on drying, ash value, microbial load, density, compressibility index and angle of repose.

**Fourier Transform Infrared (FTIR) Spectroscopy**
FTIR spectra were recorded on samples prepared in potassium bromide (KBr) disks using a Shimadzu FTIR instrument. Samples were prepared in KBr disks by means of a hydrostatic press at 6-8 tons pressure. The scanning range was 500 to 4500 cm\(^{-1}\).

**Viscosity Determination**
One (1) g of dried and finely powdered fenugreek mucilage was suspended in 75 ml of distilled water for 5 h. Distilled water added up to 100 ml to produce the concentration of 1% w/v. The mixture was homogenized by mechanical stirrer for 2 h and its viscosity determined using a Brookfield viscometer, at 37°C.

Viscosity of suspending agent
\[ \eta_1 = \eta_2 \times \left( \frac{\rho_1 t_1}{\rho_2 t_2} \right) \]

Determination of Swelling Index [11]

**Method I**
The swelling index is the volume in ml occupied by 1g of drug; including any adhering mucilage after it has been swollen in an aqueous liquid for 4h. The swelling index of Fenugreek mucilage powder, was determined according to the BP method. One gram of mucilage powder was taken in a 25 ml ground glass stoppered cylinder graduated over a height of 120 to 130 mm in 0.5 divisions. To this 25 ml of water was added and this was shaken vigorously every 10 m for 1h and then allowed to stand for 24 h. The volume occupied by mucilage was measured. The Swelling index was calculated from the mean of three determinations.

**Method II**
The natural suspending agent 1g was taken in a China dish and then 10 ml of distilled water was added and the mixture was shaken and allowed to stand for 1 hour. After 1 hour the remaining water in China dish was discarded and the weight increase of the natural suspending agent was rated.

Swelling Index % (SI) = \((W_2 – W_1/W_1) \times 100 \) \(---------- 1\)
\( W_1= \) Weight of powder at time ‘0’
\( W_2= \) Weight of powder at time ‘t’

**Solubility**
Solubility of isolated mucilage was studied using different types of solvents like water, alcohol, acetone, Polyethylene Glycols, Propylene Glycol, Glycerin, Sorbitol, Ethyl Alcohol, Methanol, Benzyl Alcohol and Isopropyl Alcohol.
Hausner ratio
Hausner calculated as the ratio of bulk density to tapped density. The Hausner ratio values less than 1.25 indicate good flow, while the values greater than 1.5 will show poor flow.

Angle of repose
The flow ability of the powder was simply measured by determination of the angle of repose (θ) using fixed funnel method. The mucoadhesive powder was allowed to flow through the funnel onto the graph paper. The radius (r) and the height (h) of the cone formed on the paper permitted the determination of the angle of repose using eq.

\[
\tan(\theta) = \frac{h}{r}.
\]

Determination of density
The bulk density (Vb) was determined by filling 50 g granules into a graduated cylinder and calculating the ratio of the sample weight to sample volume. The tapped density (Vt) was determined as the ratio of the sample weight to the final sample volume.

Result and discussion
Determination of percentage yield
After drying mucilage weight = 90 g (practical yield)
Theoretical Yield = 250 g
% Yield = Practical Yield * 100 / Theoretical Yield
% Yield = 90 * 100 / 250 = 36.

Fourier Transform Infrared (FTIR) Spectroscopy
FTIR spectra were recorded on samples prepared in potassium bromide (KBr) disks using a Shimadzu FTIR instrument. Samples were prepared in KBr disks by means of a hydrostatic press at 6-8 tons pressure. The scanning range was 500 to 4500 cm\(^{-1}\).

The absence of sharp peak at 1700–1800 cm\(^{-1}\) in the FTIR spectrum indicates that there is no carboxyl group in the extracted sample. On the other hand, the presence of peak at 1000–1200 cm\(^{-1}\) corresponds to the presence of alcoholic group mostly secondary alcohols. These findings proved that there were no uronic sugars or esters in the structure (Figure FTIR).
**Determination of angle of repose**
The mucoadhesive powder was allowed to flow through the funnel onto the graph paper. The radius (r) and the height (h) of the cone formed on the paper permitted the determination of the angle of repose using eq.

\[ \tan(\theta) = \frac{h}{r} \]

\[ h = 1.7 \text{ cm} \]
\[ r = 3.03 \]

Angle of repose \(\theta = 29.29^\circ\).

on the basis of relationship between angle of repose \(\theta\) and powder flow. The angle of repose was found to good (29.29).

**Swelling index**
Swelling index was determined was two methods. And the observation’s was presented as. Methods I – swelling index was found to be 9 ± 0.2.

Method II
Swelling Index % (SI) = \((W_2 - W_1)/W_1 \times 100\) (1)

\[ W_1 = \text{Weight of powder at time ‘}0’ \]
\[ W_2 = \text{Weight of powder at time ‘}t’ \]
\[ W_1 = 10 \text{ g}, \quad W_2 = 26, \quad \text{SI}\% = 160. \]

Result shows that the time increase, swelling index was increased, because weight gain by mucilage was proportional to rate of hydration. The direct relationship was observed between swelling index and mucilage concentration, as mucilage concentration increase swelling index increased.

**Determination of pH and viscosity**
pH of 1% solution was found to be 6.23 ± 0.2. And viscosity was observed 39 cp at 37°C (1% solution).

**Physico-chemical characterization of mucilage**

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Results</th>
</tr>
</thead>
<tbody>
<tr>
<td>Colour</td>
<td>Creamy yellowish</td>
</tr>
<tr>
<td>Taste</td>
<td>Tasteless</td>
</tr>
<tr>
<td>Odor</td>
<td>Odorless</td>
</tr>
<tr>
<td>State</td>
<td>Powder</td>
</tr>
<tr>
<td>Identification</td>
<td></td>
</tr>
<tr>
<td>a. Mounted in 96% ethanol</td>
<td>Transparent angular masses</td>
</tr>
<tr>
<td>b. Mounted in Ruthenium red</td>
<td>Particles stained red</td>
</tr>
<tr>
<td>c. Mounted in iodine solution</td>
<td>Particles stained blue</td>
</tr>
<tr>
<td>Test for carbohydrate (Mollish’s test)</td>
<td>+</td>
</tr>
<tr>
<td>Test for tannins (Ferric chloride test)</td>
<td>-</td>
</tr>
<tr>
<td>Test for chloride (silver nitrate test)</td>
<td>-</td>
</tr>
<tr>
<td>Test for sulphate (Barium chloride test)</td>
<td>-</td>
</tr>
<tr>
<td>Total bacterial count</td>
<td></td>
</tr>
<tr>
<td><em>E.coli</em></td>
<td>Absent</td>
</tr>
<tr>
<td><em>S.aureus</em></td>
<td>Absent</td>
</tr>
<tr>
<td><em>Salmonella typhi</em></td>
<td>Absent</td>
</tr>
</tbody>
</table>
### Parameters vs Results

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Results</th>
</tr>
</thead>
<tbody>
<tr>
<td>Solubility in water</td>
<td>Soluble in water (cold less, hot more)</td>
</tr>
<tr>
<td>Angle of repose</td>
<td>29.29º</td>
</tr>
<tr>
<td>Bulk density</td>
<td>0.47</td>
</tr>
<tr>
<td>Tapped density</td>
<td>0.52</td>
</tr>
<tr>
<td>pH</td>
<td>6.23 ± 0.2</td>
</tr>
<tr>
<td>True density</td>
<td>1.4g/dl</td>
</tr>
<tr>
<td>Viscosity (1%)(37ºC)</td>
<td>39 cp</td>
</tr>
<tr>
<td>Hausner ratio</td>
<td>1.10</td>
</tr>
<tr>
<td>Swelling Index</td>
<td>9 ± 0.2 (Method I); 160 (Method II)</td>
</tr>
</tbody>
</table>

### Conclusion

The extracted mucilagenous substance of Trigonellafoenumgraecum is edible, has the potential as a mucoadhesive agent even at lower concentration (1-3% w/v). So fenugreekmucoadhesive is use in nasal gel formulations with low toxicity, high viscosity. So in future fenugreek mucoadhesive agent will use in nasal gel formulations with various drugs.

### Reference